

INTERNATIONAL ORGANIZATION FOR STANDARDIZATION MEXATION OPPAHUSALUN DO CTAHDAPTUSALUNOORGANISATION INTERNATIONALE DE NORMALISATION

ANSI Internet Con South

Light olefins for industrial use — Determination of molecular oxygen in gaseous phase — Electrochemical method with a membrane-covered cell

Oléfines légères à usage industriel — Dosage de l'oxygène moléculaire en phase gazeuse — Méthode électrochimique avec une cellule à membrane

First edition - 1986-11-15

UDC 661.725 : 543.5 : 546.21

Ref. No. ISO 8173-1986 (E)

Descriptors : industrial products, olefinic hydrocarbons, chemical analysis, determination of content, oxygen, apparatus, electrochemical methods.

Price based on 5 pages

Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the technical committees are circulated to the member bodies for approval before their acceptance as International Standards by the ISO Council. They are approved in accordance with ISO procedures requiring at least 75 % approval by the member bodies voting.

International Standard ISO 8173 was prepared by Technical Committee ISO/TC 47, *Chemistry*.

Users should note that all International Standards undergo revision from time to time and that any reference made herein to any other International Standard implies its latest edition, unless otherwise stated.

© International Organization for Standardization, 1986 •

Printed in Switzerland

Light olefins for industrial use — Determination of molecular oxygen in gaseous phase — Electrochemical method with a membrane-covered cell

0 Introduction

For the determination of traces of molecular oxygen in gaseous olefins, only an electrochemical method with aqueous electrolyte at ambient temperature may be used.

The method by measurement of the paramagnetic susceptibility of the gaseous phase cannot be used on account of lack of sensitivity in the range of contents near 1 ml/m³.

The method by measurement of the electromotive force of a concentration cell with a solid zirconia electrolyte must not be used because of the temperature requirements of 700 to 850 °C.

1 Scope and field of application

This International Standard specifies an electrochemical method for the determination of molecular oxygen in light olefins in the gaseous phase using a membrane-covered electrochemical cell containing a gelled aqueous electrolyte.

The method is applicable to light olefins having an oxygen content greater than 1 ml/m³.

2 References

ISO 6349, Gas analysis – Preparation of calibration gas mixtures – Permeation method.

ISO 6377, Light olefins for industrial use — Determination of impurities by gas chromatography — General considerations.

ISO 6711, Gas analysis — Checking of calibration gas mixtures by a comparison method.

ISO 7382, Ethylene for industrial use — Sampling in the liquid and the gaseous phase.

ISO 8563, Propylene and butadiene for industrial use - Sampling in liquid phase.

3 Principle

After sampling and preparation of the laboratory sample, passage of the gaseous test portion at a given rate through an electrochemical cell previously calibrated with air or with standard mixtures containing a known amount of oxygen, then measurement of the concentration of oxygen in the gaseous olefin.

NOTE — The measurement of the concentration of the oxygen is based on the following principle:

Specific electrochemical reduction of molecular oxygen after diffusion through a polymer membrane covering the porous cathode.

The overall electric current of the galvanic cell is a function of the diffusion process.

The electric current is proportional to the partial pressure of the oxygen in the adjacent gaseous phase, i.e. it is proportional to the oxygen concentration in the gaseous phase at constant total pressure.

4 Reactions

The electrochemical reactions are as follows:

Cathode reaction

$$O_2 + 2H_2O + 4e^- \rightarrow 4OH^-$$

Anode reaction

 $4OH^- + 2Pb \rightarrow 2PbO + 2H_2O + 4e^-$

Overall reaction

 $2Pb + O_2 \rightarrow 2PbO$

5 Materials

5.1 Water, deoxygenated and deionized.

5.2 Nitrogen, having an oxygen content less than 5 ml/m³.